

# Metal Forming

A news letter dedicated to Forging Industry

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Dear Fellow Engineers,

Our group activity at Arkey Technical Training Institute & Foundry and Forging Technology Forum are concentrating on the upcoming conference on "TOMORROW'S FORGING INDUSTRY" The due date is approaching faster for September 15<sup>th</sup> & 16<sup>th</sup> 2022 at The Pride Hotel, Pune. I am sure you must have earmarked these dates in your calendar for this mega event, being held after two years, all live, off line for all of us to meet and greet each other for the first time after COVID.

Learning never stops for the wise men and I am sure to have more personal interactions with the new and old guys, net working for the benefit of all, for technological improvements in our own industries. Brochures will come as hard copies /soft copies, well in advance for you to form groups from your companies, who would attend this programme.

We are completing article on "COOLING INTENSITY OF INVERSE SOLUBILITY POLYALKYLENE GLYKOL POLYMERS AND SOME RESULTS OF INVESTIGATIONS FOCUSED ON MINIMIZING DISTORTION OF METAL COMPONENTS" which we started in the last issue. Hope you would have read it liked it. Please give us your feed back for us to know your opinion and what we should change to for topics/ discussions.

With Best Regards,

Dr. V. V. Kanetkar - Editor

FOR YOU



Copper is red, Aluminium is silver colour, Steel is Gray, even then all these look red at their respective hot working temperatures, True/ False and why?

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FORGING TECHNOLOGY FORUM

Save The Dates

15<sup>th</sup>-16<sup>th</sup> (THU-FRI) SEPTEMBER 2022

- Venue -

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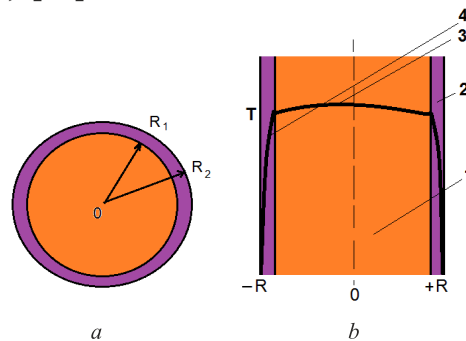
# COOLING INTENSITY OF INVERSE SOLUBILITY POLYALKYLENE GLYKOL POLYMERS AND SOME RESULTS OF INVESTIGATIONS FOCUSED ON MINIMIZING DISTORTION OF METAL COMPONENTS

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To be continued...

When generalized Biot number  $Bi_V \rightarrow 0$ , the ratio  $\frac{\bar{T}_{sf} - T_c}{\bar{T}_V - T_c} \rightarrow 1$ . It means that  $\bar{T}_{sf} \approx \bar{T}_V$ . Here  $\bar{T}_{sf}$  – average surface temperature created by radius  $R_1$  (**Fig. 5, a**);  $\bar{T}_V$  – average volume temperature. Exact profile of temperature distribution in metal can be calculated using solution (7). In this particular case  $T_c = T_s$ ,  $T_s$  is saturation temperature. On the other hand, on the surface created by radius  $R_2$ , generalized Biot number  $Bi_V \rightarrow \infty$  because during nucleate boiling process HTC reaches 200,000 W/m<sup>2</sup>K and more [10]. It means that ratio  $\frac{\bar{T}_{sf} - T_c}{\bar{T}_V - T_c} \rightarrow 0$  and  $\bar{T}_{sf} \rightarrow \bar{T}_s$ . Exact profile of temperature distribution in the insulating layer can be calculated using solution (8). Thus, insulating layer increases surface temperature of metallic sample that can compete with the quenching under pressure (**Fig. 3**) [10].



**Fig. 3** Section of a coated cylindrical probe and typical temperature distribution during quenching in polymer water solution of inverse solubility: a – section of a cylinder; b – temperature distribution in cylinder; 1 – heated metal, 2 – coating, 3 – temperature gradient in metal, 4 – temperature gradient in polymeric surface layer. The possibility of delaying of martensite transformation combined with accelerated cooling in the inverse solubility polymers is used for implementation of the new patented technologies [14]

which include:

- performing two steps quenching with cryogenic treatment on the second step to provide super-strengthening of material and increase its wear resistance [15];
- use for austempering process to obtain nanobainitic microstructure [16–18];
- performing low temperature thermomechanical treatment to obtain high strength and ductility of a

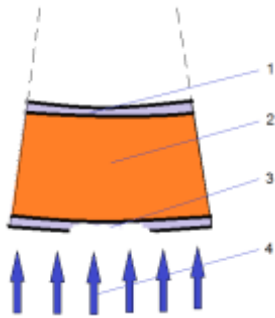
material [10].

These ideas will be further developed and discussed in the future.

## 4. Why does a big distortion and crack formation during hardening in PAG solutions sometimes takes place?

Ideal cooling in PAG solution is created by insulating layer that can govern effectively quenching processes. However, insulating layer is dissolved when surface temperature drops below 74 oC. Nobody could imagine that this fact is critical and can lead to big distortion and crack formation during quenching. During batch cooling steel parts at the bottom of load are cooled locally by water flow that creates local dissolving of polymeric layer (**Fig. 4**). At this dissolving areas martensite transformation starts first while in other areas martensitic transformation is delayed due to reason considered above. These local non-uniform martensitic transformations eliminate

completely all benefits of PAG inverse solubility polymers. The problems concerning non-uniformity of cooling were considered by authors [19, 20]. To solve this unpleasant problem, the dissolving of insulating layer must be protected at the end of hardening process by developing correct recipes of cooling. Unfortunately, heat treating industry uses only cooling curves and cooling rates of standard probe that cannot allow engineers performing correct calculations. To fix situation, a simple method of calculation is proposed on the basis of regular thermal condition theory [9, 10] and examples of calculation for complicated steel parts are provided below.



**Fig. 4.** A scheme explaining why a big distortion takes place during quenching in water PAG polymers solutions: 1 – polymer coating; 2 – quenched steel part; 3 – locally dissolved by water flow polymer coating; 4 – water flow

### 5. Examples of recipes calculations

Cooling time calculation of a core of any shape of steel part can be performed using generalized equation (10), [10]:

$$\tau = \left[ \frac{kBi_v}{2.095 + 3.867Bi_v} + \ln \left( \frac{T_0 - T_m}{T - T_m} \right) \right] \frac{K}{aKn}. \quad (10)$$

#### Example 1

Shafts 30 mm in diameter made of AISI 4140 steel are quenched in batches from 850°C in 20 % water polymer UCON™ E Quenchant at 30°C. Calculate cooling time when core temperature of shaft approaches 350°C. Form factor K for long shaft is calculated from equation  $K = R^2/5.783 = 38.9 \times 10^{-6} \text{m}^2$ . Average thermal diffusivity of super-cooled austenite is equal to  $5.5 \times 10^{-6} \text{m}^2/\text{s}$ .

Dimensionless number Kn for UCON™ E Quenchant is evaluated from **Fig. 1** and is equal to 0.327 and generalized Biot number  $Bi_v$ , according to [10] 0.45.

Cooling time is calculated using Eq. (10):

$$\tau = \left[ 0.235 + \ln \left( \frac{850^\circ\text{C} - 30^\circ\text{C}}{350^\circ\text{C} - 30^\circ\text{C}} \right) \right] \frac{38.9 \times 10^{-6} \text{m}^2}{5.5 \times 10^{-6} \text{m}^2/\text{s} \times 0.327} = 25.4 \text{ sec.}$$

After 25 seconds of cooling agitation in quench tank was stopped and as a result distortion satisfied requirements.

#### Example 2

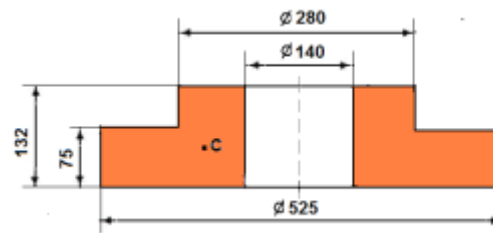
Large plugs made of AISI 5140 steel (**Fig. 5**) are batch quenched from 850°C in 1% water agitated PAG solution at 27°C. Calculate cooling time when temperature at area C reaches 400°C.

Dimensionless number Kn for agitated water solution is 0.8. Average thermal diffusivity of material is  $5.36 \times 10^{-6} \text{m}^2/\text{s}$ . Form factor K for plug is  $608 \times 10^{-6} \text{m}^2$ .

Using Eq. (10), one can calculate required cooling time for plug

$$\tau = \left[ 0.48 + \ln \left( \frac{850^\circ\text{C} - 27^\circ\text{C}}{400^\circ\text{C} - 27^\circ\text{C}} \right) \right] \frac{608 \times 10^{-6} \text{m}^2}{5.36 \times 10^{-6} \text{m}^2/\text{s} \times 0.8} = 180 \text{ sec.}$$

At a time 180 seconds agitation of solution was turned off and cooling of plug was continued to bath temperature.



**Fig. 5.** Drawing of a plug made of high quality alloy steels and used in energy manufacturing stations

### 6. Discussion

Inverse solubility of PAG polymers, used as quenchant, are tested by Inconel 600 standard probe [4] and currently there is no enough experimental data to perform correct calculations. Author believes that obtained data on effective dimensionless numbers Kn will make PAG solutions ideal quenchant for many high-tech contemporary technologies. For example, low concentration of PAG polymer solutions can be used for performing intensive quenching processes [17–19]. Small concentration (about 1 %) can eliminate completely film boiling that makes process of cooling intensive since real heat transfer coefficients during nucleate boiling are huge (**Table 1**)

[20].

**Table 1**

Influence of the PAG concentration on the duration of film- and nucleate boiling during quenching silver spherical probe 20 mm in diameter (20°C) [20].

Concentration, %	$\tau_{FB}$ , s	$\tau_{NB}$ , s	$v_{max}$ , °C/s
Water	7	2	231
PAG, 1%	0.2	2.1	674
3	1.5	2.5	467
10	1.8	4	336
20	2.6	6	251

Small concentration creates very thin surface polymeric layer that cannot radically decrease cooling rate of steel parts, however can change radically situation by decreasing initial heat flux density and such a way lowering it below the first critical heat flux density. In this case, instead of film boiling (300–500 W/m<sup>2</sup>K) will be nucleate boiling (200,000 W/m<sup>2</sup>K) that provides uniform and intensive cooling. As known, intensification of cooling doesn't affect distortion [10] if quenching is uniform. Investigation of effect of insulating layers on intensity of cooling is a new direction in heat treating industry since it makes processes cheaper, more effective and very simple for implementation.

However, prior to be widely used in the practice, additional investigations should be performed concerning kinetics of creation and dissolving polymeric surface layers during quenching.

Also, thermal properties of polymeric surface layers should be carefully investigated versus temperature and variation of thickness of insulating layers on the surface of steel parts during quenching should be known. Furthermore, Inconel 600 probe was used for comparison purposes and it cannot be used for evaluating real heat transfer coefficients which are needed for temperature field calculations and residual stresses forming. The best way of such investigation is Liscic probe combined with the video and thermal acoustic technique [21, 22]. Further detail investigations of the cooling characteristics of inverse solubility polymers can bring the great benefits for heat treating industry. To have a great success, the software governing cooling processes in PAG polymers should be developed which includes

DATABASE of form factors K for different steel parts, intensity of cooling Kn for different quenchants and thermal properties of materials to design quickly and correctly recipes for quenching processes preventing crack formation and minimizing distortion. To solve completely the problem, emitters were proposed to use during batch quenching which don't dissolve locally polymeric coatings [23]. And also contemporary approaches, hyperbolic heat conductivity equations taking into account insulating surface layers, should be used to investigate accurately further quenching processes in polymers of inverse solubility [24–28]. Especially, results of investigations are of great value connected with the forms approaching the real steel components [29, 30].

## 7. Conclusions

1. It is shown that during quenching in PAG inverse solubility polymers small temperature gradient is formed in hardened metal and maximal temperature gradient is established in the insulating polymeric layer. This fact opens the new horizons for developing high-tech processes.
2. It has been discovered that local dissolving of insulating surface polymeric layer leads to big distortion of steel parts during quenching in spite of ideal cooling in PAG polymers.
3. To reduce distortion and prevent crack formation during quenching in polymers of inverse solubility it is proposed to interrupt cooling when core temperature approaches martensite start temperature Ms or turn off agitation.
4. The data obtained by testing Inconel 600 probe can be recalculated and used for calculating cooling time for any configuration of steel part if film boiling is absent and intensity of cooling is evaluated by effective dimensionless number Kn.
5. It has been established that low water concentration of PAG polymers (about 1%) completely eliminates film boiling even during testing of silver 20 mm spherical silver probe. It creates big opportunity for intensive quenching processes, namely for IQ-2 process, during batch quenching [20].
7. Author proposes using emitters instead of propellers during batch quenching to eliminate completely local dissolving of PAG polymer layers which are a reason for big distortion of metal components [23].

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